

SECTION OVERVIEW

Unit III: Chemical Equilibria

UC Davis Chemistry Chemistry 110C

Physical Chemistry III: Thermodynamics, Equilibria and Kinetics

[Chem 110A](#) • [Chem 110B](#) • [Chem 110C](#) • **[Agenda](#)** • **[Homework](#)** • **[Worksheet Activities](#)**

Sorry, but all current instructors at UCD use the commercial textbook for this course.

Chemical equilibrium is the state in which both reactants and products are present in concentrations which have no further tendency to change with time. Usually, this state results when the forward reaction proceeds at the same rate as the reverse reaction.

[26A: Thermodynamic Description of Chemical Equilibria](#)

[26B: Statistic Description of Chemical Equilibria](#)

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