

16.11 Elementary Processes (Video)

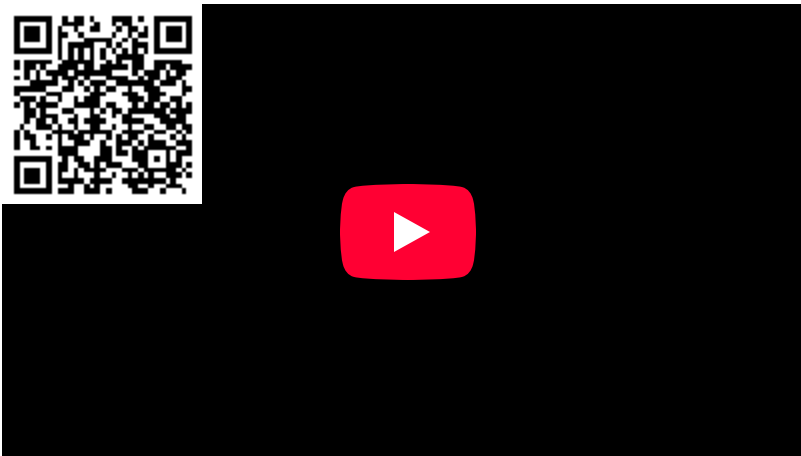
This project was preformed to supply **Libretext authors** with videos on General Chemistry topics which can be used to enhance their projects. Also, these videos are meant to act as a learning resource for **all General Chemistry students**.

Video Topics

Why do rate laws not follow stoichiometry? Because reactions often proceed by a series of simple steps called elementary processes. Reaction mechanism: The sequence of simple steps (elementary processes) by which an overall reaction occurs. For an elementary process the rate law is given by the stoichiometry.

Link to Video

Elementary Processes: <https://youtu.be/tzC7a4oxdgg>



Attribution

- Prof. Steven Farmer ([Sonoma State University](#))

16.11 Elementary Processes (Video) is shared under a [not declared](#) license and was authored, remixed, and/or curated by LibreTexts.