

14.2 Intermolecular Forces (Video)

This project was preformed to supply **Libretext Authors** with videos on General Chemistry topics which can be used to enhance their projects. Also, these videos are meant to act as a learning resource for **all General Chemistry students**.

Video Topics

Liquids and solids are held by cohesive forces between molecules. These forces are called intermolecular forces (IMF's). IMF's affect the physical properties of a compound. Solids are greater than Liquids are greater than Gases. As the IMF's increase the boiling point and melting point of a compound tend to increase. There are four main types of IMF's: Ionic, H-bonding, dipole, and dispersion

Link to Video

Intermolecular Forces: <https://youtu.be/nDUqGjcqQME>



Attribution

- Prof. Steven Farmer ([Sonoma State University](#))

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