

22.1 Spontaneity (Video)

This project was preformed to supply **Libretext Authors** with videos on General Chemistry topics which can be used to enhance their projects. Also, these videos are meant to act as a learning resource for **all General Chemistry students**.

Video Topics

Spontaneous process: A process that once started no external action is needed to make the process continue.

Non-spontaneous process: A process, which will not occur unless an external action is continuously applied.

If a process is spontaneous, the reverse process is non-spontaneous. Reactions can be spontaneous or non-spontaneous. It was originally though that a process which gives off energy (exothermic) was spontaneous. Because there are spontaneous endothermic reactions spontaneity must not entirely determined by enthalpy H .

Link to Video

Spontaneity: <https://youtu.be/Z2WjDU-LAq4>



Attribution

- Prof. Steven Farmer ([Sonoma State University](#))

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