

19.7.4 pH at the Equivalence Point in a Strong Acid/Strong Base Titration (Video)

This project was preformed to supply **Libretext authors** with videos on General Chemistry topics which can be used to enhance their projects. Also, these videos are meant to act as a learning resource for **all General Chemistry students**.

Video Topics

This video contains an example where the pH at the equivalence point of a strong acid / strong base titration is calculated.

You start with 50.0 mL of a 0.100 M solution of HCl.

Calculate the pH after the addition of 10 mL of a 0.100 M NaOH solution:

Link to Video

pH at the Equivalence Point in a Strong Acid/Strong Base Titration: <https://youtu.be/7POGDA5Ql2M>



Attribution

- Prof. Steven Farmer ([Sonoma State University](#))

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