

## CHAPTER OVERVIEW

### 20: Solubility Products

- [20.10 Solubility of Complex Ions \(Video\)](#)
- [4.11 Determining the Net Ionic Equation for a Precipitation Reaction](#)
- [20.1 Definition of a Solubility Product \(Video\)](#)
- [20.2 Finding the Solubility of a Salt \(Video\)](#)
- [4.4 Balancing Complex Chemical Equations](#)
- [20.3 Finding K<sub>sp</sub> from Ion Concentrations \(Video\)](#)
- [20.4 The Common Ion Effect in Solubility Products \(Video\)](#)
- [20.5 Determining if a Precipitate forms \(The Ion Product\) \(Video\)](#)
- [20.6 Removal of an Ion from Solution Using Precipitation \(Video\)](#)
- [20.7 The Separation of Two Ions by a Difference in Solubility \(Video\)](#)
- [20.8 Solubility Products and pH \(Video\)](#)
- [20.9 pH and the Precipitation of Insoluble Salts \(Video\)](#)

---

[20: Solubility Products](#) is shared under a [not declared](#) license and was authored, remixed, and/or curated by LibreTexts.