

17.11 Converting K_p to K_c (Video)

This project was preformed to supply **Libretext authors** with videos on General Chemistry topics which can be used to enhance their projects. Also, these videos are meant to act as a learning resource for **all General Chemistry students**.

Video Topics

This video contains the solution to the following question: For the reaction $\text{CO}_{(g)} + \text{Cl}_{2(g)} \rightarrow \text{COCl}_{2(g)}$ If the equilibrium partial pressures at 0 °C are: 1.2×10^2 atm COCl_2 ; 5.0×10^{-4} atm CO and 4.0×10^{-4} atm Cl_2 what is K_c?

Link to Video

Converting K_p to K_c: https://youtu.be/WK_5qOEIgms



Attribution

- Prof. Steven Farmer ([Sonoma State University](#))

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