

9.9: In-Text References

1. Recall that we have used hindered bases before. In that case, it was to prevent nucleophilic attack when we wanted to bring about an E2 elimination reaction and tertiary butoxide ($\text{KOC}(\text{CH}_3)_3$), as our hindered base.
 2. Diagram from Wikipedia <https://en.Wikipedia.org/wiki/File:G...tated.svg#file> licensed under a creative commons license.
 3. In fact the name of the most common form of glucose is (2R, 3R, 4S, 5S, 6R) – 6(hydroxymethyl) tetrahydro-2H-pyran-2,3,4,5-tetraol. There is no need to remember this!
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