

SECTION OVERVIEW

1.2.1: Specific Interactions

Intermolecular forces are forces of attraction or repulsion which act between neighboring particles (atoms, molecules or ions). They are weak compared to the intramolecular forces, which keep a molecule together (e.g., covalent and ionic bonding).

Topic hierarchy

This page titled [1.2.1: Specific Interactions](#) is shared under a [not declared](#) license and was authored, remixed, and/or curated by [Layne Morsch](#).