

## SECTION OVERVIEW

### Unit 4: Equilibrium in Chemical Reactions

An General Chemistry Libretexts Textmap organized around the textbook

## Principles of Modern Chemistry

by Oxtoby, Gillis, and Campion

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Chemical equilibrium is a state in which the rate of the forward reaction equals the rate of the backward reaction. In other words, there is no net change in concentrations of reactants and products. This kind of equilibrium is also called dynamic equilibrium and can be used in a wide range of processes including both chemical and physical changes.

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