

5.4: Spherical Harmonics

The rigid rotor problem was solved using the Schrödinger equation

$$-\frac{\hbar^2}{2\mu r^2} \left(\frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \sin\theta \frac{\partial}{\partial\theta} + \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial\phi^2} \right) \psi(\theta, \phi) = E\psi(\theta, \phi)$$

As it turns out, the solutions to this equation are very important in a number of areas in chemistry and physics. The eigenfunctions are known as the **spherical harmonics** ($Y_l^{m_l}(\theta, \phi)$) and they appear in every problem that has spherical symmetry. The Spherical Harmonics satisfy the relationship

$$\left(\frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \sin\theta \frac{\partial}{\partial\theta} + \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial\phi^2} \right) Y_l^{m_l}(\theta, \phi) = \hbar^2 l(l+1) Y_l^{m_l}(\theta, \phi)$$

Each function $Y_l^{m_l}(\theta, \phi)$ has three parts: 1) a normalization constant, 2) an associated Legendre polynomial in $\cos(\theta)$, and 3) an imaginary (for $m_l \neq 0$) exponential in ϕ .

$$Y_l^{m_l}(\theta, \phi) = \left[\frac{(2l+1)(l-|m_l|)!}{4\pi(l+|m_l|)!} \right]^{\frac{1}{2}} P_l^{|m_l|}(\cos\theta) e^{im_l\phi}$$

The first few Spherical harmonics are shown in the table below.

l	m_l	$Y_l^{m_l}(\theta, \phi)$
0	0	$\sqrt{\frac{1}{4\pi}}$
1	0	$\sqrt{\frac{3}{4\pi}} \cos(\theta)$
	± 1	± 1
2	± 1	$\sqrt{\frac{5}{16\pi}} (3\cos^2(\theta) - 1)$
	0	$\sin(\theta) \cos(\theta) e^{\pm i\phi}$
	± 2	$\sqrt{\frac{15}{32\pi}} \sin^2(\theta) e^{\pm 2i\phi}$

Notice the $(2l+1)$ degeneracy in these functions, due to the $(2l+1)$ values of m_l for each value of l . Also, it is useful to not that these functions all have l angular nodes (values of θ that cause the wavefunction to vanish.) For the $l=1$ wavefunctions, these nodes occur at $\theta = \pi/2$ for $m_l = 0$ and at $\theta = 0$ for $m_l = \pm 1$. The number of nodes in each wavefunction is a useful property to know when discussing how these functions related to the radial wavefunction in the Hydrogen atom.

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