

5.11: Proton NMR problems

Objectives

- Solve unknown problems using ^1H NMR spectra and molecular formula.

Note

Helpful resources for solving these types of problems:

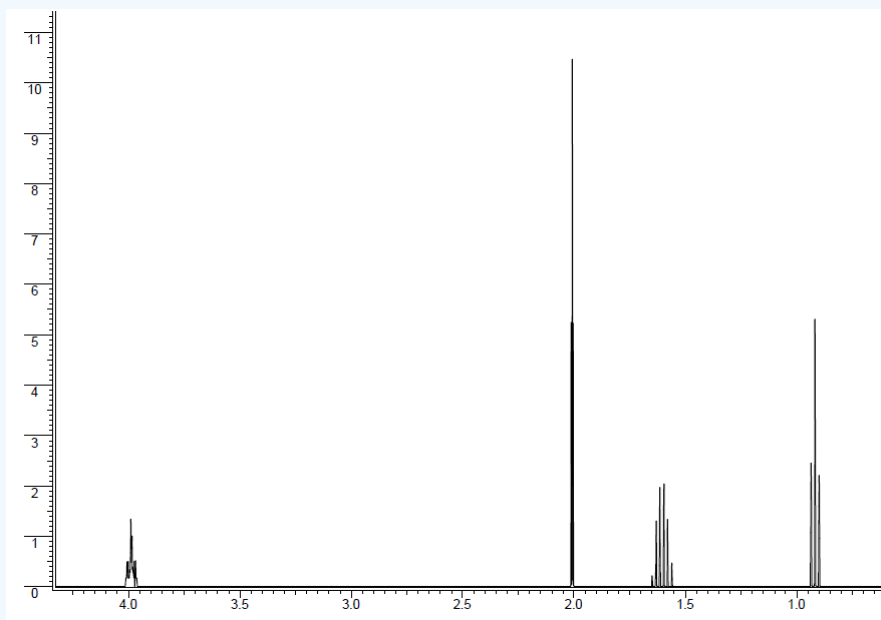
- [Degree of Unsaturation Equation](#)
- [Chemical Shift Data Table](#)
- [Coupling Constant Data Table](#)
- [IR Data Table](#)

You may also want to read through some worked problems on how to solve unknown structure determination problems ([Section 5.10](#)).

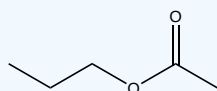
? Exercise 5.11.1

Determine the structure for the unknown molecule with the molecular formula of $\text{C}_5\text{H}_{10}\text{O}_2$.

^1H NMR: The ratio of protons is 2:3:2:3. $J = 7$ Hz for all coupling.



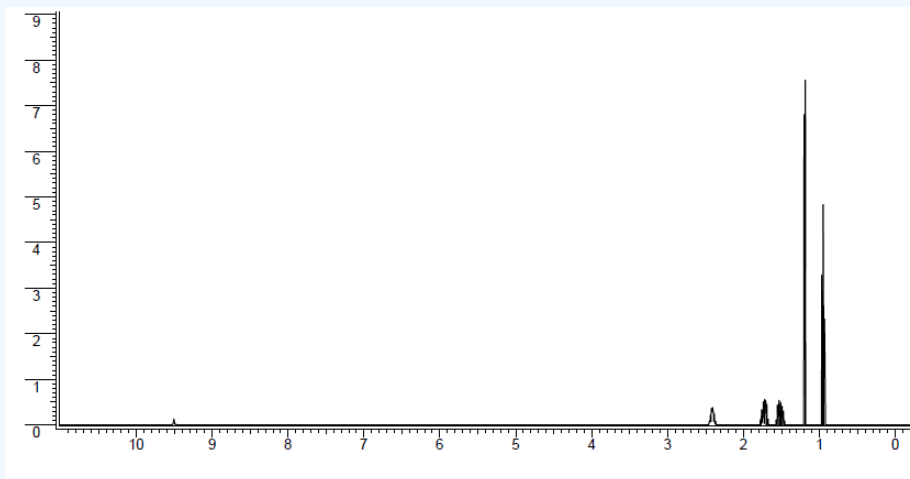
Answer



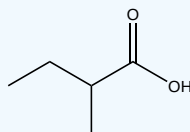
? Exercise 5.11.2

Determine the structure for the unknown molecule with the molecular formula of $C_5H_{10}O_2$.

1H NMR: The ratio of protons is 1:1:1:1:3:3. The peak at 9.5 ppm is a singlet. The peak at 2.41 ppm is sextet ($J = 7$ Hz). The peak at 1.72 ppm is a multiplet ($J = 7$ Hz, 25 Hz). The peak at 1.53 ppm is a multiplet ($J = 7$ Hz, 25 Hz). The peak at 1.20 ppm is a doublet ($J = 7$ Hz). The peak at 0.95 ppm is a triplet ($J = 7$ Hz).



Answer

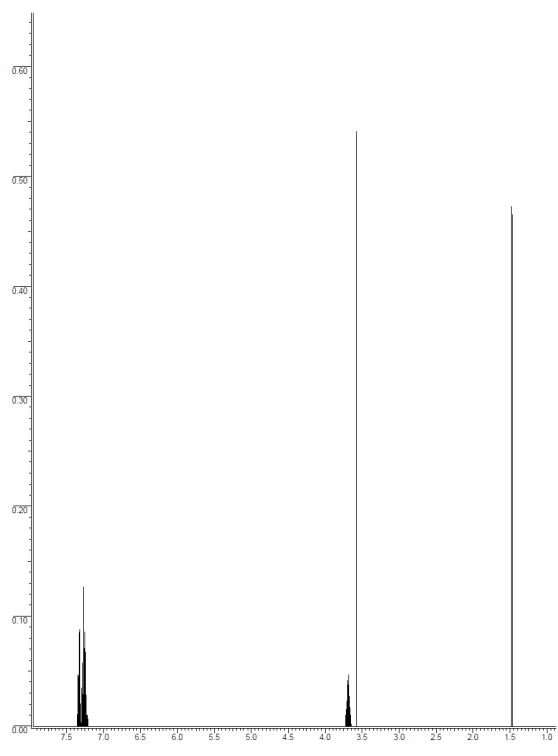


Note: The $-CH_2-$ protons are **diastereotopic**, so they show up differently in 1H NMR spectra.

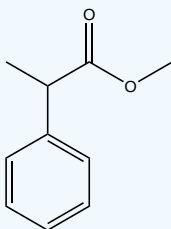
? Exercise 5.11.3

Determine the structure for the unknown molecule with the molecular formula of $C_{10}H_{12}O_2$.

1H NMR: The ratio of protons is 5:1:3:3. The peak at 7.5 ppm is a multiplet. The peak at 3.70 ppm is quartet ($J = 7$ Hz). The peak at 3.58 ppm is a singlet. The peak at 1.48 ppm is a doublet ($J = 7$ Hz).



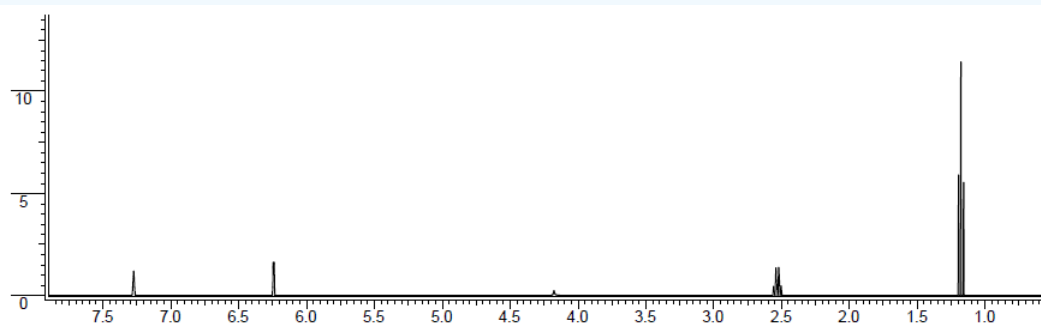
Answer



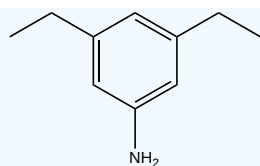
? Exercise 5.11.4

Determine the structure for the unknown molecule with the molecular formula of $C_{10}H_{15}N$.

1H NMR: The ratio of protons is 1:2:2:4:6. The peak at 7.3 ppm is a triplet ($J = 2\text{ Hz}$) and the peak at 6.24 ppm is a doublet ($J = 2\text{ Hz}$).



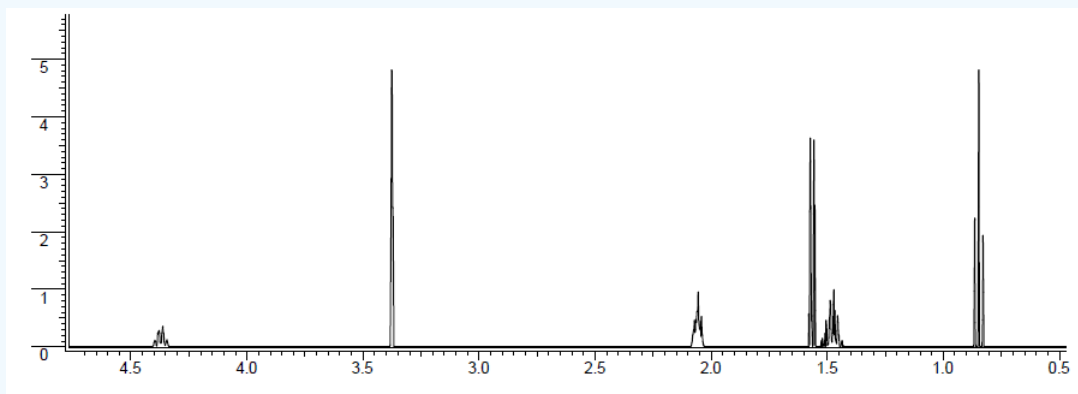
Answer



? Exercise 5.11.5

Determine the structure for the unknown molecule with the molecular formula of $C_7H_{14}O$.

1H NMR: The ratio of protons is 1:3:2:3:2:3. The peak at 4.36 ppm is a triplet of quartets ($J = 17$ Hz and 7Hz). The peak at 3.37 ppm is a singlet. The peak at 2.04 ppm is a doublet of triplets ($J = 17$ Hz and 7Hz). The peak at 1.56 ppm is a doublet ($J = 7$ Hz). The peak at 1.48 ppm is a sextet ($J = 7$ Hz). The peak at 0.85 ppm is a triplet ($J = 7$ Hz).



Answer

