

## 5.11: Proton NMR problems

### Objectives

- Solve unknown problems using  $^1\text{H}$  NMR spectra and molecular formula.

### Note

Helpful resources for solving these types of problems:

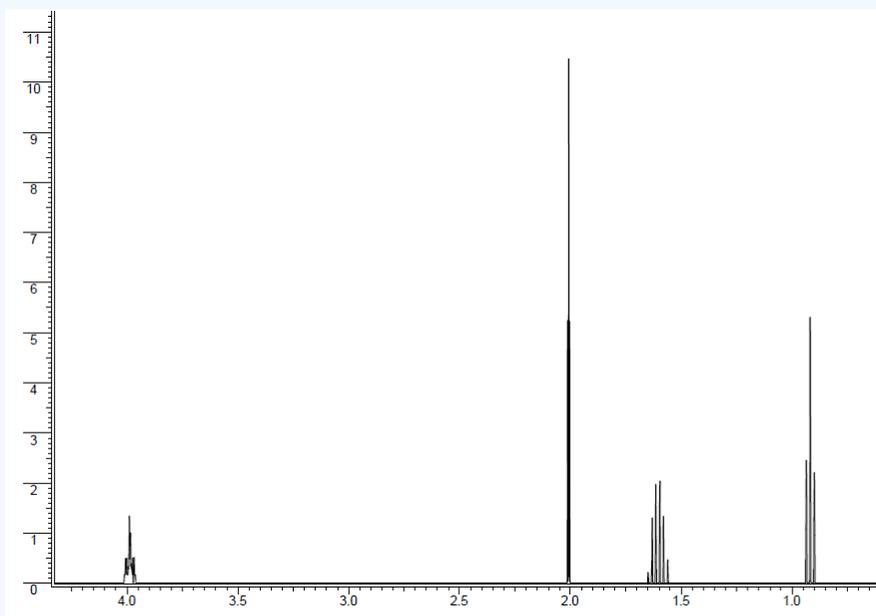
- [Degree of Unsaturation Equation](#)
- [Chemical Shift Data Table](#)
- [Coupling Constant Data Table](#)
- [IR Data Table](#)

You may also want to read through some worked problems on how to solve unknown structure determination problems ([Section 5.10](#)).

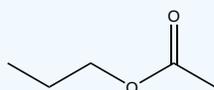
### ? Exercise 5.11.1

Determine the structure for the unknown molecule with the molecular formula of  $\text{C}_5\text{H}_{10}\text{O}_2$ .

$^1\text{H}$  NMR: The ratio of protons is 2:3:2:3.  $J = 7$  Hz for all coupling.



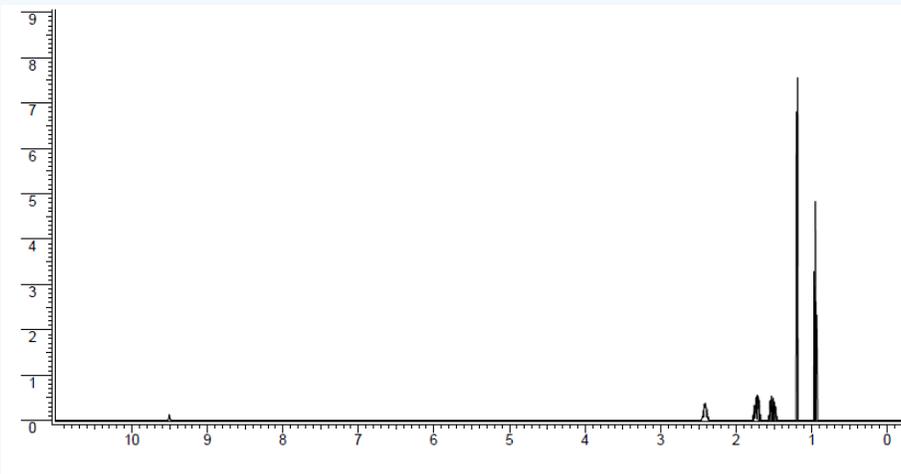
**Answer**



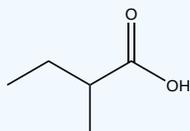
### ? Exercise 5.11.2

Determine the structure for the unknown molecule with the molecular formula of  $C_5H_{10}O_2$ .

$^1H$  NMR: The ratio of protons is 1:1:1:1:3:3. The peak at 9.5 ppm is a singlet. The peak at 2.41 ppm is sextet ( $J = 7$  Hz). The peak at 1.72 ppm is a multiplet ( $J = 7$  Hz, 25 Hz). The peak at 1.53 ppm is a multiplet ( $J = 7$  Hz, 25 Hz). The peak at 1.20 ppm is a doublet ( $J = 7$  Hz). The peak at 0.95 ppm is a triplet ( $J = 7$  Hz).



**Answer**

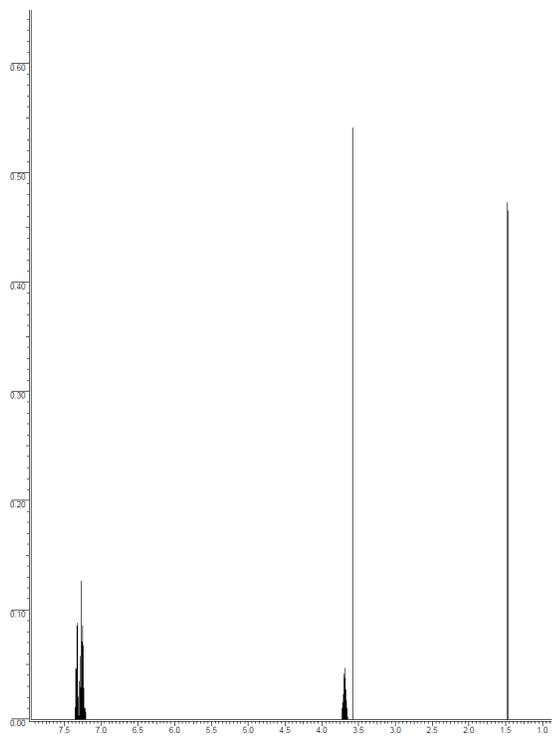


Note: The  $-CH_2-$  protons are **diastereotopic**, so they show up differently in  $^1H$  NMR spectra.

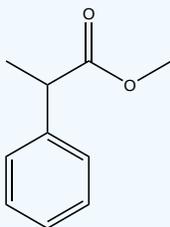
### ? Exercise 5.11.3

Determine the structure for the unknown molecule with the molecular formula of  $C_{10}H_{12}O_2$ .

$^1H$  NMR: The ratio of protons is 5:1:3:3. The peak at 7.5 ppm is a multiplet. The peak at 3.70 ppm is quartet ( $J = 7$  Hz). The peak at 3.58 ppm is a singlet. The peak at 1.48 ppm is a doublet ( $J = 7$  Hz).



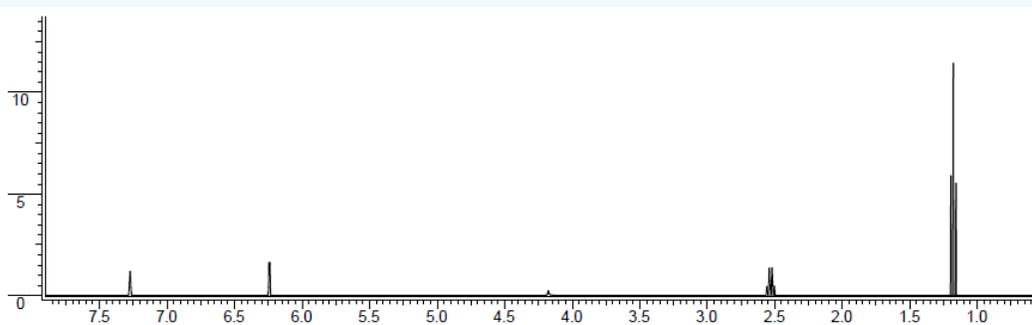
**Answer**



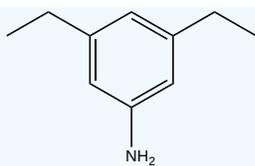
### ? Exercise 5.11.4

Determine the structure for the unknown molecule with the molecular formula of  $C_{10}H_{15}N$ .

$^1H$  NMR: The ratio of protons is 1:2:2:4:6. The peak at 7.3 ppm is a triplet ( $J = 2\text{Hz}$ ) and the peak at 6.24 ppm is a doublet ( $J = 2\text{ Hz}$ ).



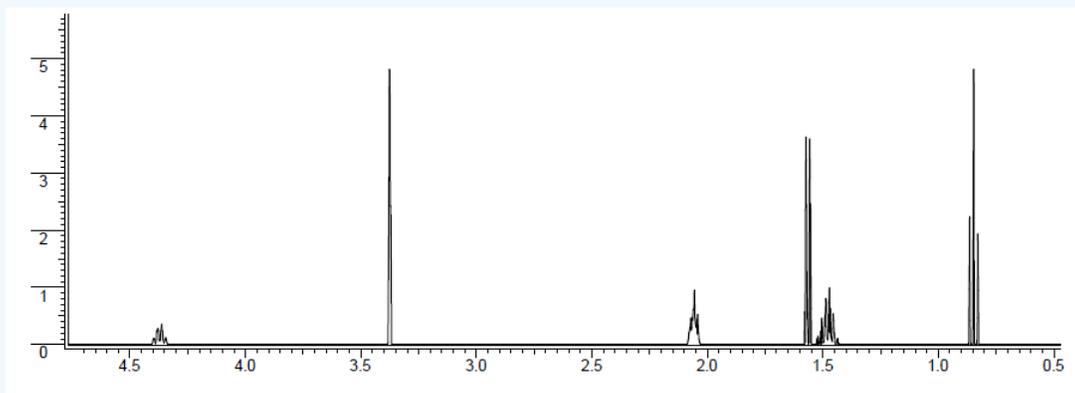
**Answer**



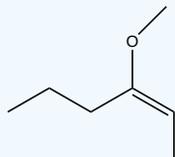
### ? Exercise 5.11.5

Determine the structure for the unknown molecule with the molecular formula of  $C_7H_{14}O$ .

$^1H$  NMR: The ratio of protons is 1:3:2:3:2:3. The peak at 4.36 ppm is a triplet of quartets ( $J = 17$  Hz and 7Hz). The peak at 3.37 ppm is a singlet. The peak at 2.04 ppm is a doublet of triplets ( $J = 17$  Hz and 7Hz). The peak at 1.56 ppm is a doublet ( $J = 7$  Hz). The peak at 1.48 ppm is a sextet ( $J = 7$  Hz). The peak at 0.85 ppm is a triplet ( $J = 7$  Hz).



**Answer**



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