

## Detailed Licensing

### Overview

**Title:** CHEM 110: Bellingham Technical College

**Webpages:** 66

**Applicable Restrictions:** Noncommercial

**All licenses found:**

- [CC BY-NC-SA 3.0](#): 80.3% (53 pages)
- [Undeclared](#): 19.7% (13 pages)

### By Page

- CHEM 110: Bellingham Technical College - *Undeclared*
  - Front Matter - *Undeclared*
    - TitlePage - *Undeclared*
    - InfoPage - *Undeclared*
    - Table of Contents - *Undeclared*
    - Licensing - *Undeclared*
  - 1: Module 1 - *Undeclared*
    - 1.1: Prelude to Chemistry - *CC BY-NC-SA 3.0*
    - 1.2: Basic Definitions - *CC BY-NC-SA 3.0*
    - 1.3: Chemistry as a Science - *CC BY-NC-SA 3.0*
    - 1.4: Prelude to Measurements - *CC BY-NC-SA 3.0*
    - 1.5: Expressing Numbers - *CC BY-NC-SA 3.0*
    - 1.6: Expressing Units - *CC BY-NC-SA 3.0*
    - 1.7: Significant Figures - *CC BY-NC-SA 3.0*
    - 1.8: Converting Units - *CC BY-NC-SA 3.0*
    - 1.9: Other Units - Temperature and Density - *CC BY-NC-SA 3.0*
  - 2: Module 2 - *CC BY-NC-SA 3.0*
    - 2.1: Prelude to Atoms, Molecules, and Ions - *CC BY-NC-SA 3.0*
    - 2.2: Atomic Theory - *CC BY-NC-SA 3.0*
    - 2.3: Organization of Electrons in Atoms - *CC BY-NC-SA 3.0*
    - 2.4: Electronic Structure and the Periodic Table - *CC BY-NC-SA 3.0*
    - 2.5: Molecules and Chemical Nomenclature - *CC BY-NC-SA 3.0*
    - 2.6: Masses of Atoms and Molecules - *CC BY-NC-SA 3.0*
    - 2.7: Ions and Ionic Compounds - *CC BY-NC-SA 3.0*
    - 2.8: Prelude to Chemical Bonds - *CC BY-NC-SA 3.0*
    - 2.9: Lewis Electron Dot Diagrams - *CC BY-NC-SA 3.0*
    - 2.10: Electron Transfer - Ionic Bonds - *CC BY-NC-SA 3.0*
    - 2.11: Covalent Bonds - *CC BY-NC-SA 3.0*
    - 2.12: Other Aspects of Covalent Bonds - *CC BY-NC-SA 3.0*
  - 3: Module 3 - *CC BY-NC-SA 3.0*
    - 3.1: Prelude to Organic Chemistry - *CC BY-NC-SA 3.0*
    - 3.2: Hydrocarbons - *CC BY-NC-SA 3.0*
    - 3.3: Branched Hydrocarbons - *CC BY-NC-SA 3.0*
    - 3.4: Alkyl Halides and Alcohols - *CC BY-NC-SA 3.0*
    - 3.5: Other Oxygen-Containing Functional Groups - *CC BY-NC-SA 3.0*
    - 3.6: Other Functional Groups - *CC BY-NC-SA 3.0*
    - 3.7: Polymers - *CC BY-NC-SA 3.0*
    - 3.8: Prelude to Chemical Reactions - *CC BY-NC-SA 3.0*
    - 3.9: The Chemical Equation - *CC BY-NC-SA 3.0*
    - 3.10: Types of Chemical Reactions - Single and Double Replacement Reactions - *CC BY-NC-SA 3.0*
    - 3.11: Ionic Equations - A Closer Look - *CC BY-NC-SA 3.0*
    - 3.12: Composition, Decomposition, and Combustion Reactions - *CC BY-NC-SA 3.0*
    - 3.13: Neutralization Reactions - *CC BY-NC-SA 3.0*
    - 3.14: Oxidation-Reduction Reactions - *Undeclared*
  - 4: Module 4 - *CC BY-NC-SA 3.0*
    - 4.1: Stoichiometry - *CC BY-NC-SA 3.0*
    - 4.2: The Mole - *CC BY-NC-SA 3.0*
    - 4.3: The Mole in Chemical Reactions - *CC BY-NC-SA 3.0*
    - 4.4: Mole-Mass and Mass-Mass Calculations - *CC BY-NC-SA 3.0*
    - 4.5: Yields - *CC BY-NC-SA 3.0*
    - 4.6: Limiting Reagents - *CC BY-NC-SA 3.0*
    - 4.7: Energy - *CC BY-NC-SA 3.0*
    - 4.8: Work and Heat - *CC BY-NC-SA 3.0*
    - 4.9: Enthalpy and Chemical Reactions - *CC BY-NC-SA 3.0*
    - 4.10: Intermolecular Forces - *CC BY-NC-SA 3.0*

- 4.11: Phase Transitions - Melting, Boiling, and Subliming - *CC BY-NC-SA 3.0*
- 4.12: Prelude to Solutions - *CC BY-NC-SA 3.0*
- 4.13: Definitions - *CC BY-NC-SA 3.0*
- 4.14: Quantitative Units of Concentration - *CC BY-NC-SA 3.0*
- 4.15: Dilutions and Concentrations - *CC BY-NC-SA 3.0*
- 4.16: Concentrations as Conversion Factors - *CC BY-NC-SA 3.0*
- Back Matter - *Undeclared*
  - Index - *Undeclared*
  - Glossary - *Undeclared*
  - Detailed Licensing - *Undeclared*
  - Detailed Licensing - *Undeclared*