

## 5.2: Pressure and Work Virial Theorems

As noted earlier, the quantity  $-\partial H/\partial V$  is a measure of the instantaneous value of the internal pressure  $P_{\text{int}}$ . Let us look at the ensemble average of this quantity

$$\begin{aligned}\langle P_{\text{int}} \rangle &= -\frac{1}{\Delta} C_N \int_0^\infty dV e^{-\beta PV} \int dx \frac{\partial H}{\partial V} e^{-\beta H(x)} \\ &= \frac{1}{\Delta} C_N \int_0^\infty dV e^{-\beta PV} \int dx kT \frac{\partial}{\partial V} e^{-\beta H(x)} \\ &= \frac{1}{\Delta} \int_0^\infty dV e^{-\beta PV} kT \frac{\partial}{\partial V} Q(N, V, T)\end{aligned}$$

Doing the volume integration by parts gives

$$\begin{aligned}\langle P_{\text{int}} \rangle &= \frac{1}{\Delta} [e^{-\beta PV} kT Q(N, V, T)] \Big|_0^\infty - \frac{1}{\Delta} \int_0^\infty dV kT \left( \frac{\partial}{\partial V} e^{-\beta PV} \right) Q(N, V, T) \\ &= P \frac{1}{\Delta} \int_0^\infty dV e^{-\beta PV} Q(N, V, T) \\ &= P\end{aligned}$$

Thus,

$$\langle P_{\text{int}} \rangle = P$$

This result is known as the pressure virial theorem. It illustrates that the average of the quantity  $-\partial H/\partial V$  gives the fixed pressure  $P$  that defines the ensemble. Another important result comes from considering the ensemble average  $-\partial H/\partial V$

$$\langle P_{\text{int}} V \rangle = \frac{1}{\Delta} \int_0^\infty dV e^{-\beta PV} kT V \frac{\partial}{\partial V} Q(N, V, T)$$

Once again, integrating by parts with respect to the volume yields

$$\begin{aligned}\langle P_{\text{int}} V \rangle &= \frac{1}{\Delta} [e^{-\beta PV} kT V Q(N, V, T)] \Big|_0^\infty - \frac{1}{\Delta} \int_0^\infty dV kT \left( \frac{\partial}{\partial V} V e^{-\beta PV} \right) Q(N, V, T) \\ &= \frac{1}{\Delta} \left[ -kT \int_0^\infty dV e^{-\beta PV} Q(V) + P \int_0^\infty dV e^{-\beta PV} V Q(V) \right] \\ &= -kT + P \langle V \rangle\end{aligned}$$

or

$$\langle P_{\text{int}} V \rangle + kT = P \langle V \rangle$$

This result is known as the **work virial theorem**. It expresses the fact that equipartitioning of energy also applies to the volume degrees of freedom, since the volume is now a fluctuating quantity.

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