

10.9: Momentum-Space Variation Method for the Quartic Oscillator

For unit mass the quartic oscillator has the following energy operator in atomic units in coordinate space.

$$H = -\frac{1}{2} \frac{d^2}{dx^2}$$

Suggested trial wavefunction:

$$\psi(x, \beta) = \left(\frac{2\beta}{\pi}\right)^{\frac{1}{4}} \exp(-\beta x^2)$$

Demonstrate that the wavefunction is normalized.

$$\int_{-\infty}^{\infty} \psi(x, \beta)^2 dx \text{ assume, } \beta > 0 \rightarrow 1$$

Fourier transform the coordinate wavefunction into the momentum representation.

$$\Phi(p, \beta) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} \exp(-ipx) \psi(x, \beta) dx \xrightarrow[\text{simplify}]{\text{assume, } \beta > 0} \frac{1}{2} \frac{2^{\frac{3}{4}}}{\pi^{\frac{1}{4}}} \frac{e^{-\frac{1}{4} \frac{p^2}{\beta}}}{\beta^{\frac{1}{4}}}$$

Demonstrate that the momentum wavefunction is normalized.

$$\int_{-\infty}^{\infty} \overline{\Phi(p, \beta)} \Phi(p, \beta) dp \text{ assume, } \beta > 0 \rightarrow 1$$

The quartic oscillator energy operator in momentum space:

$$H = \frac{p^2}{2} + \frac{d^4}{dp^4}$$

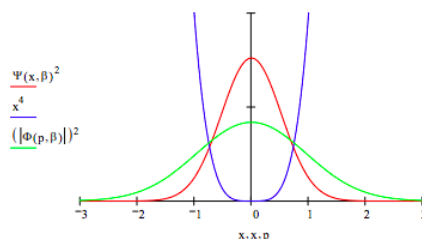
Evaluate the variational energy integral.

$$E(\beta) = \int_{-\infty}^{\infty} \overline{\Phi(p, \beta)} \frac{p^2}{2} \Phi(p, \beta) dp + \int_{-\infty}^{\infty} \overline{\Phi(p, \beta)} \frac{d^4}{dp^4} \Phi(p, \beta) dp \xrightarrow[\text{simplify}]{\text{assume, } \beta > 0} \frac{1}{16} \frac{8\beta^3 + 3}{\beta^2}$$

Minimize the energy with respect to the variational parameter β and report its optimum value and the ground-state energy.

$$\beta = 1 \quad \beta = \text{Minimize}(E, \beta) \quad \beta = 0.90856 \quad E(\beta) = 0.688142$$

Plot the coordinate and momentum wavefunctions and the potential energy on the same graph.



These results demonstrate the uncertainty principle. For the harmonic potential, $x^2/2$, the coordinate and momentum wavefunctions are identical. Compared to the harmonic potential the quartic potential, x^4 , constrains the spatial wavefunction leading to less uncertainty in position. The uncertainty principle, therefore, requires an increase in the momentum uncertainty. This is clearly revealed in graph above.

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