

9.12: Numerical Solutions for Morse Oscillator

Schrödinger's equation is integrated numerically for the first three energy states for the Morse oscillator. The integration algorithm is taken from J. C. Hansen, *J. Chem. Educ. Software*, **8C2**, 1996.

Set parameters:

$$n = 300$$

$$x_{\min} = -2$$

$$x_{\max} = 12$$

$$\Delta = \frac{x_{\max} - x_{\min}}{n - 1}$$

$$\mu = 1$$

$$D = 2$$

$$\beta = 2$$

$$x_e = 0$$

Calculate position vector, the potential energy matrix, and the kinetic energy matrix. Then combine them into a total energy matrix.

$$i = 1 \dots n \quad j = 1 \dots n \quad x_i = x_{\min} + (i - 1) \Delta$$

$$V_{i,j} = \text{if} \left[i = j, D[1 - \exp[\beta(x_i - x_e)]]^2, 0 \right]$$

$$T_{i,j} = \text{if} \left[i = j, \frac{\pi^2}{6\mu\Delta^2}, \frac{(-1)^{i-j}}{(i-j)^2\mu\Delta^2} \right]$$

Hamiltonian matrix: $H = T + V$

Find eigenvalues: $E = \text{sort}(\text{eigenvals}(H))$

Display three eigenvalues: $m = 1 \dots 3$

$E_m =$

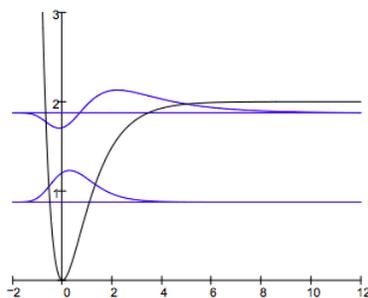
| |
|--------|
| 0.8750 |
| 1.8750 |
| 2.0596 |

Calculate associated eigenfunctions:

$$k = 1 \dots 3$$

$$\psi(k) = \text{eigenvec}(H, E_k)$$

Plot the potential energy and selected eigenfunctions:



For $V = ax^n$, the virial theorem requires the following relationship between the expectation values for kinetic and potential energy:

$$\langle T \rangle = 0.5n \langle V \rangle.$$

The calculations below show that virial theorem is not satisfied for the Morse oscillator. The reason is revealed in the following series expansion in x . The expansion contains cubic, quartic and higher order terms in x , so the virial theorem **does not** apply to the quartic oscillator.

$D(1 - \exp(-\beta x))^2$ converts to the series $D\beta^2 x^2 + (-D)\beta^3 x^3 + \frac{7}{12}D\beta^4 x^4 + O(x^5)$

$$\begin{array}{l}
 \left(\begin{array}{ccc}
 \text{'' Kinetic Energy ''} & \text{'' Potential Energy ''} & \text{'' Total Energy ''} \\
 \psi(1)^T T \psi(1) & \psi(1)^T V \psi(1) & E_1 \\
 \psi(2)^T T \psi(2) & \psi(2)^T V \psi(2) & E_2
 \end{array} \right) \\
 = \left(\begin{array}{ccc}
 \text{'' Kinetic Energy ''} & \text{'' Potential Energy ''} & \text{'' Total Energy ''} \\
 0.3750 & 0.5000 & 0.8750 \\
 0.3754 & 1.4996 & 1.8750
 \end{array} \right)
 \end{array}$$

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