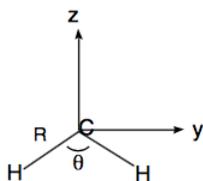


22.4.4: iv. Problem Solutions

Q1



Molecule I Molecule II (22.4.4.1)

$$R_{CH} = 1.121\text{\AA} \quad R_{CH} = 1.076\text{\AA} \quad (22.4.4.2)$$

$$\angle_{HCH} = 104^\circ \quad \angle_{HCH} = 136^\circ \quad (22.4.4.3)$$

$$y_H = R \sin\left(\frac{\theta}{2}\right) = \pm 0.8834 \quad y_H = \pm 0.9976 \quad (22.4.4.4)$$

$$z_H = R \cos\left(\frac{\theta}{2}\right) = -0.2902 \quad z_H = -0.4031 \quad (22.4.4.5)$$

Center of Mass(COM): clearly, X = Y = 0, (22.4.4.6)

$$Z = \frac{12(0) - 2R \cos\left(\frac{\theta}{2}\right)}{14} = -0.0986 \quad Z = -0.0576 \quad (22.4.4.7)$$

a.

$$I_{xx} = \sum_j m_j (y_j^2 + z_j^2) - M(Y^2 + Z^2) \quad (22.4.4.8)$$

$$T_{xy} = -\sum_j m_j x_j y_j - MXY \quad (22.4.4.9)$$

$$I_{xx} = 2(1.121)^2 - 14(-0.0986)^2 \quad I_{xx} = 2(1.076)^2 - 14(-0.0576)^2 \quad (22.4.4.10)$$

$$= 2.377 \quad = 2.269 \quad (22.4.4.11)$$

$$I_{yy} = 2(0.6902)^2 - 14(-0.0986)^2 \quad I_{yy} = 2(0.4031)^2 - 14(-0.0576)^2 \quad (22.4.4.12)$$

$$= 0.8167 \quad = 0.2786 \quad (22.4.4.13)$$

$$I_{zz} = 2(0.8834)^2 \quad I_{zz} = 2(0.9976)^2 \quad (22.4.4.14)$$

$$= 1.561 \quad = 1.990 \quad (22.4.4.15)$$

$$I_{xz} = I_{yz} = I_{xy} = 0 \quad (22.4.4.16)$$

b. Since the moment of inertia tensor is already diagonal, the principal moments of inertia have already been determined to be

$$(I_a \langle I_b \langle I_c) : \quad (22.4.4.17)$$

$$I_{yy} \langle I_{zz} \langle I_{xx} \quad I_{yy} \langle I_{zz} \langle I_{xx} \quad (22.4.4.18)$$

$$0.8167 \langle 1.561 \langle 2.377 \quad 0.2786 \langle 1.990 \langle 2.269 \quad (22.4.4.19)$$

$$\text{Using the formula: } A = \frac{h}{8\pi^2 c I_a} = \frac{6.62 \times 10^{-27}}{8\pi^2 (3 \times 10^{10}) I_a}$$

$$A = \frac{16.84}{I_a} \text{cm}^{-1}$$

similarly, $B = \frac{16.84}{I_b} \text{cm}^{-1}$, and $C = \frac{16.84}{I_c} \text{cm}^{-1}$.

So,

$$\text{Molecule I} \qquad \text{Molecule II} \qquad (22.4.4.20)$$

$$y \Rightarrow A = 20.62 \qquad y \Rightarrow A = 60.45 \qquad (22.4.4.21)$$

$$z \Rightarrow B = 10.79 \qquad z \Rightarrow B = 8.46 \qquad (22.4.4.22)$$

$$x \Rightarrow C = 7.08 \qquad x \Rightarrow C = 7.42 \qquad (22.4.4.23)$$

c. Averaging B + C:

$$B = \frac{B+C}{2} = 8.94 \qquad B = \frac{B+C}{2} = 7.94 \qquad (22.4.4.24)$$

$$A - B = 11.68 \qquad A - B = 52.51 \qquad (22.4.4.25)$$

Using the Prolate top formula

$$E = (A - B)K^2 + BJ(J + 1),$$

$$\text{Molecule I} \qquad \text{Molecule II} \qquad (22.4.4.26)$$

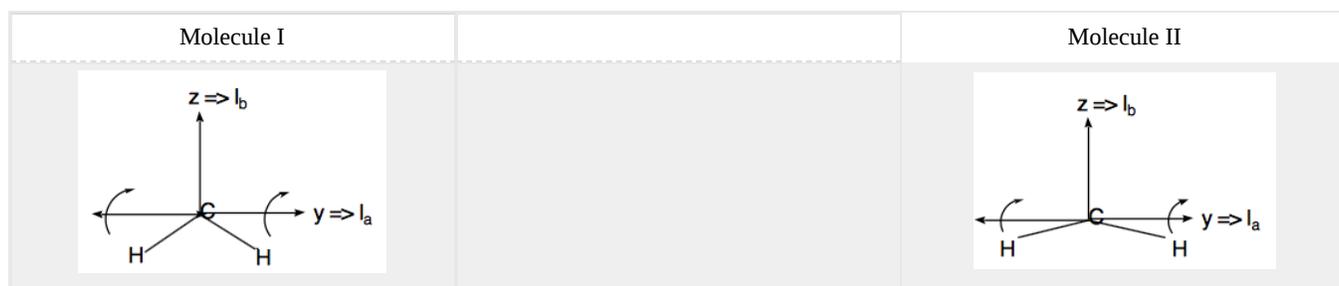
$$E = 11.68K^2 + 8.94J(J + 1) \qquad E = 52.51K^2 + 7.94J(J + 1) \qquad (22.4.4.27)$$

Levels: $J = 0, 1, 2, \dots$ and $K = 0, 1, \dots, J$

For a given level defined by J and K , there are M_J degeneracies given by: $(2J + 1) \times$

$$\begin{cases} 1 \text{ for } K = 0 \\ 2 \text{ for } K \neq 0 \end{cases} \qquad (22.4.4.28)$$

d.



e. Since $\vec{\mu}$ is along Y, $\Delta K = 0$ since K describes rotation about the y axis.

Therefore $\Delta J = \pm 1$

f. Assume molecule I is CH_2^- and molecule II is CH_2 . Then, $\Delta E = E_{J_j}(\text{CH}_2)$, where:

$E(\text{CH}_2) = 52.51K^2 + 7.94J(J + 1)$, and $E(\text{CH}_2^-) = 11.68K^2 + 8.94J(J + 1)$

For R-branches: $J_j = J_i + 1, \Delta K = 0$;

$$\begin{aligned}
 \Delta E_R &= E_{J_j}(CH_2) - E_{J_i}(CH_2) & (22.4.4.29) \\
 &= 7.94(J_i + 1)(J_i + 1 + 1) - 8.94J_i(J_i + 1) & (22.4.4.30) \\
 &= (J_i + 1)\{7.94(J_i + 1 + 1) - 8.94J_i\} & (22.4.4.31) \\
 &= (J_i + 1)\{(7.94 - 8.94)J_i + 2(7.94)\} & (22.4.4.32) \\
 &= (J_i + 1)\{-J_i + 15.88\} & (22.4.4.33)
 \end{aligned}$$

For P-branches: $J_j - 1, \Delta K = 0$;

$$\begin{aligned}
 \Delta E_P &= E_{J_j}(CH_2) - E_{J_i}(CH_2) & (22.4.4.34) \\
 &= 7.94(J_i - 1)(J_i - 1 + 1) - 8.94J_i(J_i + 1) & (22.4.4.35) \\
 &= J_i\{7.94(J_i - 1) - 8.94(J_i + 1)\} & (22.4.4.36) \\
 &= J_i\{(7.94 - 8.94)J_i - 7.94 - 8.94\} & (22.4.4.37) \\
 &= J_i\{-J_i - 16.88\} & (22.4.4.38)
 \end{aligned}$$

This indicates that the R branch lines occur at energies which grow closer and closer together as J increases (since the $15.88 - J_i$ term will cancel). The P branch lines occur at energies which lie more and more negative (i.e. to the left of the origin). So, you can predict that if molecule I is CH_2^- and molecule II is CH_2 then the R-branch has a band head and the P-branch does not. This is observed therefore our assumption was correct:

molecule I is CH_2^- and molecule II is CH_2 .

g. The band head occurs when $\frac{d(\Delta E_R)}{dJ} = 0$.

$$\frac{d(\Delta E_R)}{dJ} = \frac{d}{dJ} [(J_i + 1)\{-J_i + 15.88\}] = 0 \quad (22.4.4.39)$$

$$= \frac{d}{dJ} (-J_i^2 - J_i + 15.88J_i + 15.88) = 0 \quad (22.4.4.40)$$

$$= -2J_i + 14.88 = 0 \quad (22.4.4.41)$$

$$\therefore J_i = 7.44, \text{ so } J = 7 \text{ or } 8. \quad (22.4.4.42)$$

At $J = 7.44$:

$$\Delta E_R = (J + 1)\{-J + 15.88\} \quad (22.4.4.43)$$

$$\Delta E_R = (7.44 + 1)\{-7.44 + 15.88\} = (8.44)(8.44) = 71.2\text{cm}^{-1} \text{ above the origin.} \quad (22.4.4.44)$$

Q2

a.

D_{6h}	E	$2C_6$	$2C_3$	C_2	$3C_2'$	$3C_2''$	i	$2S_3$	$2S_6$	σ_h	$3\sigma_d$	$3\sigma_v$		
A_{1g}	1	1	1	1	1	1	1	1	1	1	1	1	R_z	x^2+y^2, z^2
A_{2g}	1	1	1	1	-1	-1	1	1	1	1	-1	-1		
B_{1g}	1	-1	1	-1	1	-1	1	-1	1	-1	1	-1		
B_{2g}	1	-1	1	-1	-1	1	1	-1	1	-1	-1	1	(R_x, R_y)	(xz, yz) (x^2-y^2, xy)
E_{1g}	2	1	-1	-2	0	0	2	1	-1	-2	0	0		
E_{2g}	2	-1	-1	2	0	0	2	-1	-1	2	0	0	z	
A_{1u}	1	1	1	1	1	1	-1	-1	-1	-1	-1	-1		
A_{2u}	1	1	1	1	-1	-1	-1	-1	-1	-1	1	1		
B_{1u}	1	-1	1	-1	1	-1	-1	1	-1	1	-1	1	(x, y)	
B_{2u}	1	-1	1	-1	-1	1	-1	1	-1	1	1	-1		
E_{1u}	2	1	-1	-2	0	0	-2	-1	1	2	0	0		
E_{2u}	2	-1	-1	2	0	0	-2	1	1	-2	0	0		
Γ_{C-H}	6	0	0	0	0	2	0	0	0	6	2	0		

b. The number of irreducible representations may be found by using the following formula:

$$n_{irrep} = \frac{1}{g} \sum_R \xi_{red}(R) \xi_{irrep}(R), \quad (22.4.4.45)$$

where $g =$ the order of the point group (24 for D_{6h}). (22.4.4.46)

$$n_{A_{1g}} = \frac{1}{24} \sum_R \Gamma_{C-H}(R) A_{1g}(R) \quad (22.4.4.47)$$

$$= \frac{1}{24} \{ (1)(6)(1) + (2)(0)(1) + (2)(0)(1) + (1)(0)(1) + \dots \quad (22.4.4.48)$$

$$+ (3)(0)(1) + (3)(2)(1) + (1)(0)(1) + (2)(0)(1) + \dots \quad (22.4.4.49)$$

$$+ (2)(0)(1) + (1)(6)(1) + (3)(2)(1) + (3)(0)(1) \} \quad (22.4.4.50)$$

$$= 1 \quad (22.4.4.51)$$

$$n_{A_{2g}} = \frac{1}{24} \{ (1)(6)(1) + (2)(0)(1) + (2)(0)(1) + (1)(0)(1) + \dots \quad (22.4.4.52)$$

$$+ (3)(0)(-1) + (3)(2)(-1) + (1)(0)(1) + (2)(0)(1) + \dots \quad (22.4.4.53)$$

$$+ (2)(0)(1) + (1)(6)(1) + (3)(2)(-1) + (3)(0)(-1) \} \quad (22.4.4.54)$$

$$= 0 \quad (22.4.4.55)$$

$$n_{B_{1g}} = \frac{1}{24} \{ (1)(6)(1) + (2)(0)(-1) + (2)(0)(1) + (1)(0)(-1) + \dots \quad (22.4.4.56)$$

$$+ (3)(0)(1) + (3)(2)(-1) + (1)(0)(1) + (2)(0)(-1) + \dots \quad (22.4.4.57)$$

$$+ (2)(0)(1) + (1)(6)(-1) + (3)(2)(1) + (3)(0)(-1) \} \quad (22.4.4.58)$$

$$= 0 \quad (22.4.4.59)$$

$$n_{B_{2g}} = \frac{1}{24} \{ (1)(6)(1) + (2)(0)(-1) + (2)(0)(1) + (1)(0)(-1) + \dots \quad (22.4.4.60)$$

$$+ (3)(0)(-1) + (3)(2)(1) + (1)(0)(1) + (2)(0)(-1) + \dots \quad (22.4.4.61)$$

$$+ (2)(0)(1) + (1)(6)(-1) + (3)(2)(-1) + (3)(0)(1) \} \quad (22.4.4.62)$$

$$= 0 \quad (22.4.4.63)$$

$$n_{E_{1g}} = \frac{1}{24} \{ (1)(6)(2) + (2)(0)(1) + (2)(0)(-1) + (1)(0)(-2) + \dots \quad (22.4.4.64)$$

$$+ (3)(0)(0) + (3)(2)(0) + (1)(0)(2) + (2)(0)(1) + \dots \quad (22.4.4.65)$$

$$+ (2)(0)(-1) + (1)(6)(-2) + (3)(2)(0) + (3)(0)(0) \} \quad (22.4.4.66)$$

$$= 0 \quad (22.4.4.67)$$

$$n_{E_{2g}} = \frac{1}{24} \{ (1)(6)(2) + (2)(0)(-1) + (2)(0)(-1) + (1)(0)(2) + \dots \quad (22.4.4.68)$$

$$+ (3)(0)(0) + (3)(2)(0) + (1)(0)(2) + (2)(0)(-1) + \dots \quad (22.4.4.69)$$

$$\begin{aligned}
 & + (2)(0)(-1) + (1)(6)(2) + (3)(2)(0) + (3)(0)(0) \} & (22.4.4.70) \\
 = & 0 & (22.4.4.71) \\
 n_{A_{1u}} = & \frac{1}{24} \{ (1)(6)(1) + (2)(0)(1) + (2)(0)(1) + (1)(0)(1) + \dots & (22.4.4.72) \\
 & + (3)(0)(1) + (3)(2)(1) + (1)(0)(-1) + (2)(0)(-1) + \dots & (22.4.4.73) \\
 & + (2)(0)(-1) + (1)(6)(-1) + (3)(2)(-1) + (3)(0)(-1) \} & (22.4.4.74) \\
 = & 0 & (22.4.4.75) \\
 n_{A_{2u}} = & \frac{1}{24} \{ (1)(6)(1) + (2)(0)(1) + (2)(0)(1) + (1)(0)(1) + \dots & (22.4.4.76) \\
 & + (3)(0)(-1) + (3)(2)(-1) + (1)(0)(-1) + (2)(0)(-1) + \dots & (22.4.4.77) \\
 & + (2)(0)(-1) + (1)(6)(-1) + (3)(2)(1) + (3)(0)(1) \} & (22.4.4.78) \\
 = & 0 & (22.4.4.79) \\
 n_{B_{1u}} = & \frac{1}{24} \{ (1)(6)(1) + (2)(0)(-1) + (2)(0)(1) + (1)(0)(-1) + \dots & (22.4.4.80) \\
 & + (3)(0)(1) + (3)(2)(-1) + (1)(0)(-1) + (2)(0)(1) + \dots & (22.4.4.81) \\
 & + (2)(0)(-1) + (1)(6)(1) + (3)(2)(-1) + (3)(0)(1) \} & (22.4.4.82) \\
 = & 0 & (22.4.4.83) \\
 n_{B_{2u}} = & \frac{1}{24} \{ (1)(6)(1) + (2)(0)(-1) + (2)(0)(1) + (1)(0)(-1) + \dots & (22.4.4.84) \\
 & + (3)(0)(-1) + (3)(2)(1) + (1)(0)(-1) + (2)(0)(1) + \dots & (22.4.4.85) \\
 & + (2)(0)(-1) + (1)(6)(1) + (3)(2)(1) + (3)(0)(-1) \} & (22.4.4.86) \\
 = & 1 & (22.4.4.87) \\
 n_{E_{1u}} = & \frac{1}{24} \{ (1)(6)(2) + (2)(0)(1) + (2)(0)(-1) + (1)(0)(-2) + \dots & (22.4.4.88) \\
 & + (3)(0)(0) + (3)(2)(0) + (1)(0)(-2) + (2)(0)(-1) + \dots & (22.4.4.89) \\
 & + (2)(0)(1) + (1)(6)(2) + (3)(2)(0) + (3)(0)(0) \} & (22.4.4.90) \\
 = & 1 & (22.4.4.91) \\
 n_{E_{2u}} = & \frac{1}{24} \{ (1)(6)(2) + (2)(0)(-1) + (2)(0)(-1) + (1)(0)(2) + \dots & (22.4.4.92) \\
 & + (3)(0)(0) + (3)(2)(0) + (1)(0)(-2) + (2)(0)(1) + \dots & (22.4.4.93) \\
 & + (2)(0)(1) + (1)(6)(-2) + (3)(2)(0) + (3)(0)(0) \} & (22.4.4.94) \\
 = & 0 & (22.4.4.95)
 \end{aligned}$$

We see that $\Gamma_{C-H} = A_{1g} \oplus E_{2g} \oplus B_{2u} \oplus E_{1u}$

c. x and $y \Rightarrow E_{1u}$, $z \Rightarrow A_{2u}$, so, the ground state A_{1g} level can be excited to the degenerate E_{1u} level by coupling through the x or y transition dipoles. Therefore E_{1u} is infrared active and \perp polarized.

d. $(x^2 + y^2, z^2) \Rightarrow A_{1g}$, $(xz, yz) \Rightarrow E_{1g}$, $(x^2 - y^2, xy) \Rightarrow E_{2g}$, so, the ground state A_{1g} level can be excited to the degenerate E_{2g} level by coupling through the $x^2 - y^2$ or xy transitions or be excited to the degenerate A_{1g} level by coupling through the xz or yz transitions. Therefore A_{1g} and E_{2g} are Raman active.

Q3

a.

$$\frac{d}{dr} \left(\frac{F}{r} \right) = \frac{F'}{r} - \frac{F}{r^2} \quad (22.4.4.96)$$

$$r^2 \frac{d}{dr} \left(\frac{F}{r} \right) = rF' - F \quad (22.4.4.97)$$

$$\frac{d}{dr} \left(r^2 \frac{d}{dr} \left(\frac{F}{r} \right) \right) = F' - F' + rF'' \quad (22.4.4.98)$$

So,

$$\frac{-\hbar^2}{2\mu r^2} \frac{d}{dr} \left(r^2 \frac{d}{dr} \left(\frac{F}{r} \right) \right) = \frac{-\hbar^2}{2\mu} \frac{F''}{r} \quad (22.4.4.99)$$

Rewriting the radial Schrödinger equation with the substitution: $R = \frac{F}{r}$ gives:

$$\frac{-\hbar^2}{2\mu r^2} \frac{d}{dr} \left(r^2 \frac{d(Fr^{-1})}{dr} \right) + \frac{J(J+1)\hbar^2}{2\mu r^2} \left(\frac{F}{r} \right) + \frac{1}{2} k(r-r_e)^2 \left(\frac{F}{r} \right) = \left(\frac{F}{r} \right)$$

Using the above derived identity gives:

$$\frac{-\hbar^2}{2\mu} \frac{F''}{r} + \frac{J(J+1)\hbar^2}{2\mu r^2} \left(\frac{F}{r} \right) + \frac{1}{2} k(r-r_e)^2 \left(\frac{F}{r} \right) = E \left(\frac{F}{r} \right)$$

Cancelling out an r^{-1} :

$$\frac{-\hbar^2}{2\mu} F'' + \frac{J(J+1)\hbar^2}{2\mu r^2} F + \frac{1}{2} k(r-r_e)^2 F = EF$$

b.

$$\frac{1}{r^2} = \frac{1}{(r_e + \Delta r)^2} = \frac{1}{r_e^2 \left(1 + \frac{\Delta r}{r_e} \right)} \approx \frac{1}{r_e^2} \left(1 - \frac{2\Delta r}{r_e} + \frac{3\Delta r^2}{r_e^2} \right)$$

So,

$$\frac{J(J+1)\hbar^2}{2\mu r^2} \approx \frac{J(J+1)\hbar^2}{2\mu r_e^2} \left(1 - \frac{2\Delta r}{r_e} + \frac{3\Delta r^2}{r_e^2} \right)$$

c. Using this substitution we now have:

$$\frac{-\hbar^2}{2\mu} F'' + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \left(1 - \frac{2\Delta r}{r_e} + \frac{3\Delta r^2}{r_e^2} \right) F + \frac{1}{2} k(r-r_e)^2 F = EF$$

Now, regroup the terms which are linear and quadratic in $\Delta r = r - r_e$:

$$\begin{aligned} & \frac{1}{2} k \Delta r^2 + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{3}{r_e^2} \Delta r^2 - \frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{2}{r_e} \Delta r \\ & = \left(\frac{1}{2} k + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{3}{r_e^2} \right) \Delta r^2 - \left(\frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{2}{r_e} \right) \Delta r \end{aligned}$$

Now, we must complete the square:

$$a\Delta r^2 - b\Delta r = a\left(\Delta r - \frac{b}{2a}\right)^2 - \frac{b^2}{4a}.$$

So,

$$\left(\frac{1}{2}k + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{3}{r_e^2}\right) \left(\Delta r - \frac{\frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{1}{r_e}}{\frac{1}{2}k + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{3}{r_e^2}}\right)^2 - \frac{\left(\frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{1}{r_e}\right)^2}{\frac{1}{2}k + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \frac{3}{r_e^2}}$$

Now, redefine the first term as $\frac{1}{2}\bar{k}$, second term as $(r - \bar{r}_e)^2$, and the third term as $-\Delta$ giving:

$$\frac{1}{2}\bar{k}(r - \bar{r}_e)^2 - \Delta$$

From:

$$\frac{-\hbar^2}{2\mu} F'' + \frac{J(J+1)\hbar^2}{2\mu r_e^2} \left(1 - \frac{2\Delta r}{r_e} + \frac{3\Delta r^2}{r_e^2}\right) F + \frac{1}{2}k(r - r_e)^2 F = EF, \quad (22.4.4.100)$$

$$\frac{-\hbar^2}{2\mu} F'' + \frac{J(J+1)\hbar^2}{2\mu r_e} F + \left(\frac{J(J+1)\hbar^2}{2\mu r_e^2} \left(-\frac{2\Delta r}{r_e} + \frac{3\Delta r^2}{r_e^2}\right) + \frac{1}{2}k\Delta r^2\right) F = EF \quad (22.4.4.101)$$

and making the above substitution result in:

$$\frac{-\hbar^2}{2\mu} F'' + \frac{J(J+1)\hbar^2}{2\mu r_e^2} F + \left(\frac{1}{2}\bar{k}(r - \bar{r}_e)^2 - \Delta\right) F = EF,$$

or,

$$\frac{-\hbar^2}{2\mu} F'' + \frac{1}{2}\bar{k}(r - \bar{r}_e)^2 F = \left(E - \frac{J(J+1)\hbar^2}{2\mu r_e^2} + \Delta\right) F.$$

d. Since the above is nothing but a harmonic oscillator differential equation in x with force constant \bar{k} and equilibrium bond length \bar{r}_e , we know that:

$$\frac{-\hbar^2}{2\mu} F'' + \frac{1}{2}\bar{k}(r - \bar{r}_e)^2 F = \varepsilon F, \text{ has energy levels:} \quad (22.4.4.102)$$

$$\varepsilon = \hbar\sqrt{\frac{\bar{k}}{\mu}} \left(v + \frac{1}{2}\right), v = 0, 1, 2, \dots \quad (22.4.4.103)$$

So,

$$E + \Delta - \frac{J(J+1)\hbar^2}{2\mu r_e^2} = \varepsilon \quad (22.4.4.104)$$

tell us that:

$$E = \hbar \sqrt{\frac{\bar{k}}{\mu}} \left(v + \frac{1}{2} \right) + \frac{J(J+1)\hbar^2}{2\mu r_e^2} - \Delta. \quad (22.4.4.105)$$

As J increases, \bar{r}_e increases because of the centrifugal force pushing the two atoms apart. On the other hand \bar{k} increases which indicates that the molecule finds it more difficult to stretch against both the centrifugal and Hooke's Law (spring) Harmonic force field. The total energy level (labeled by J and v) will equal a rigid rotor component $\frac{J(J+1)\hbar^2}{2\mu r_e^2}$ plus a Harmonic oscillator part $\hbar \sqrt{\frac{\bar{k}}{\mu}} \left(v + \frac{1}{2} \right)$ (which has a force constant \bar{k} which increases with J).

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