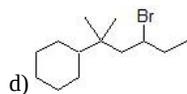
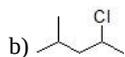


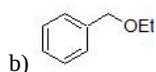
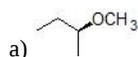
7.23: SOLUTIONS TO ADDITIONAL EXERCISES

SN2

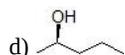
7-1



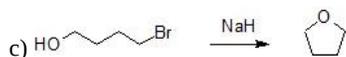
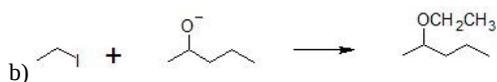
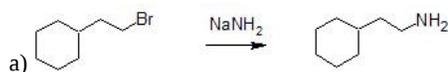
7-2



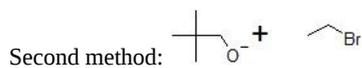
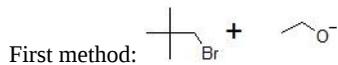
c) No reaction



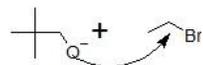
7-3



7-4



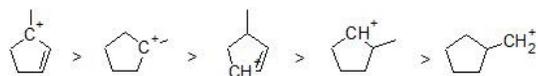
7-5 The second method is more efficient since the alkyl halide is not sterically hindered



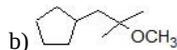
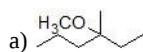
7-6 $\text{H}_2\text{O} < \text{NH}_2^- < \text{CH}_3\text{CH}_2\text{O}^- < \text{CH}_3\text{O}^-$

SN1

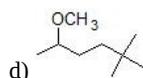
7-7



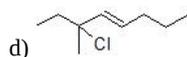
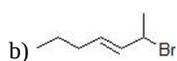
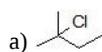
7-8



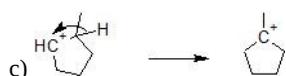
c) No reaction



7-9



7-10



SN2 VS SN1

7-11

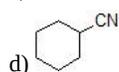
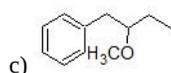
a) S_N2

b) S_N2

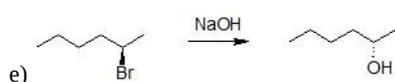
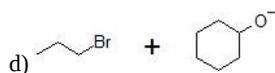
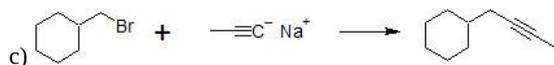
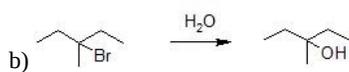
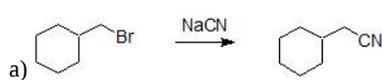
c) S_N1

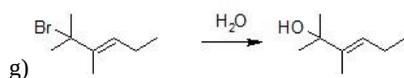
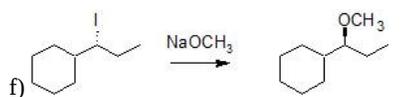
d) S_N1

7-12



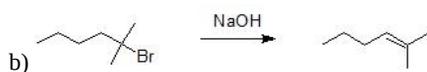
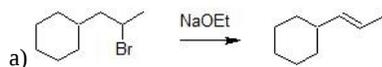
7-13



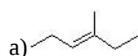


E2 VS E1

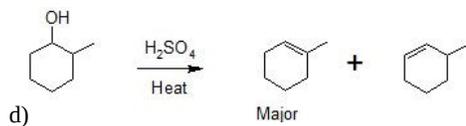
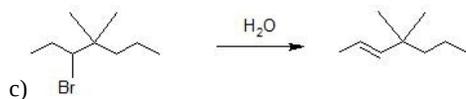
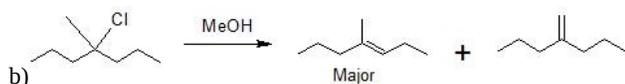
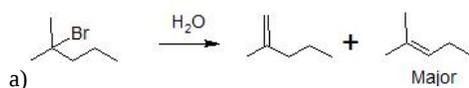
7-14



7-15



7-16



SUBSTITUTION VS ELIMINATION

7-17

a) Cl^- ; strong nucleophile

b) NaH ; strong base

c) t-BuO^- ; strong base

d) OH^- ; strong nucleophile ; strong base

e) H_2O ; weak nucleophile ; weak base

f) HS^- ; strong nucleophile

g) MeOH ; weak nucleophile ; weak base

7-18

a) $\text{E}_2, \text{S}_{\text{N}}1$

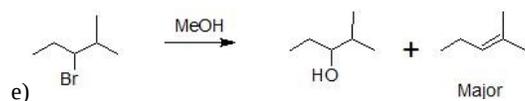
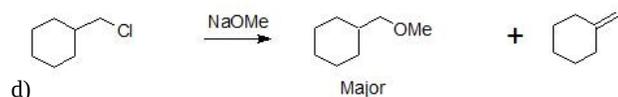
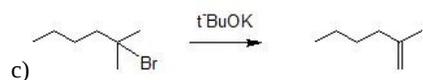
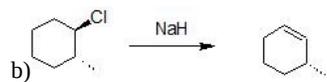
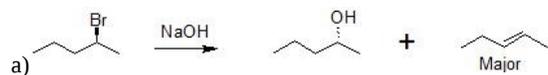
b) $\text{S}_{\text{N}}2, \text{E}_2$

c) $\text{S}_{\text{N}}2$

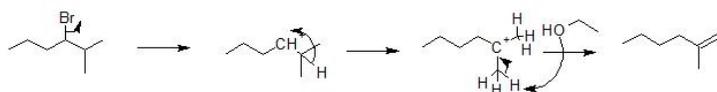
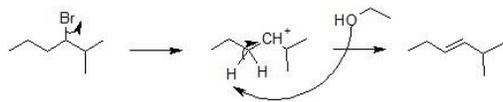
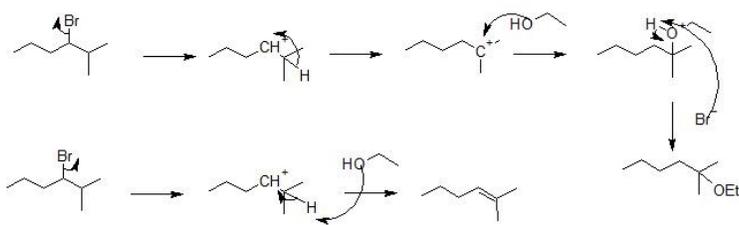
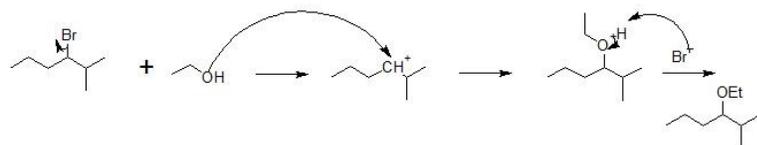
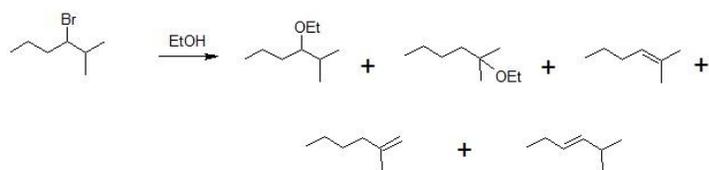
d) $\text{S}_{\text{N}}1, \text{E}_1$

e) E2, S_N1

7-19



7-20



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