

CHAPTER OVERVIEW

8: Multielectron Atoms

Electrons with more than one atom, such as Helium (He), and Nitrogen (N), are referred to as multi-electron atoms. Hydrogen is the only atom in the periodic table that has one electron in the orbitals under ground state. We will learn how additional electrons behave and affect a certain atom.

- [8.1: Atomic and Molecular Calculations are Expressed in Atomic Units](#)
- [8.2: Perturbation Theory and the Variational Method for Helium](#)
- [8.3: Hartree-Fock Equations are Solved by the Self-Consistent Field Method](#)
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