

## 10.3: Two Spin One-Half Particles

Consider a system consisting of two spin one-half particles. Suppose that the system does not possess any orbital angular momentum. Let  $\mathbf{S}_1$  and  $\mathbf{S}_2$  be the spin angular momentum operators of the first and second particles, respectively, and let

$$\mathbf{S} = \mathbf{S}_1 + \mathbf{S}_2 \quad (10.3.1)$$

be the total spin angular momentum operator. By analogy with the previous analysis, we conclude that it is possible to simultaneously measure either  $S_1^2$ ,  $S_2^2$ ,  $S^2$ , and  $S_z$ , or  $S_1^2$ ,  $S_2^2$ ,  $S_{1z}$ ,  $S_{2z}$ , and  $S_z$ . Let the quantum numbers associated with measurements of  $S_1^2$ ,  $S_{1z}$ ,  $S_2^2$ ,  $S_{2z}$ ,  $S^2$ , and  $S_z$  be  $s_1$ ,  $m_{s_1}$ ,  $s_2$ ,  $m_{s_2}$ ,  $s$ , and  $m_s$ , respectively. In other words, if the spinor  $\chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)}$  is a simultaneous eigenstate of  $S_1^2$ ,  $S_2^2$ ,  $S_{1z}$ , and  $S_{2z}$ , then

$$\begin{aligned} S_1^2 \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)} &= s_1(s_1 + 1) \hbar^2 \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)}, \\ S_2^2 \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)} &= s_2(s_2 + 1) \hbar^2 \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)}, \\ S_{1z} \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)} &= m_{s_1} \hbar \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)}, \\ S_{2z} \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)} &= m_{s_2} \hbar \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)}, \\ S_z \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)} &= m_s \hbar \chi_{s_1, s_2; m_{s_1}, m_{s_2}}^{(1)}. \end{aligned}$$

Likewise, if the spinor  $\chi_{s_1, s_2; s, m_s}^{(2)}$  is a simultaneous eigenstate of  $S_1^2$ ,  $S_2^2$ ,  $S^2$ , and  $S_z$ , then

$$\begin{aligned} S_1^2 \chi_{s_1, s_2; s, m_s}^{(2)} &= s_1(s_1 + 1) \hbar^2 \chi_{s_1, s_2; s, m_s}^{(2)}, \\ S_2^2 \chi_{s_1, s_2; s, m_s}^{(2)} &= s_2(s_2 + 1) \hbar^2 \chi_{s_1, s_2; s, m_s}^{(2)}, \\ S^2 \chi_{s_1, s_2; s, m_s}^{(2)} &= s(s + 1) \hbar^2 \chi_{s_1, s_2; s, m_s}^{(2)}, \\ S_z \chi_{s_1, s_2; s, m_s}^{(2)} &= m_s \hbar \chi_{s_1, s_2; s, m_s}^{(2)}. \end{aligned}$$

Of course, because both particles have spin one-half,  $s_1 = s_2 = 1/2$ , and  $s_{1z}, s_{2z} = \pm 1/2$ . Furthermore, by analogy with previous analysis,

$$m_s = m_{s_1} + m_{s_2}. \quad (10.3.2)$$

Now, we saw, in the previous section, that when spin  $l$  is added to spin one-half then the possible values of the total angular momentum quantum number are  $j = l \pm 1/2$ . By analogy, when spin one-half is added to spin one-half then the possible values of the total spin quantum number are  $s = 1/2 \pm 1/2$ . In other words, when two spin one-half particles are combined, we either obtain a state with overall spin  $s = 1$ , or a state with overall spin  $s = 0$ . To be more exact, there are three possible  $s = 1$  states (corresponding to  $m_s = -1, 0, 1$ ), and one possible  $s = 0$  state (corresponding to  $m_s = 0$ ). The three  $s = 1$  states are generally known as the *triplet* states, whereas the  $s = 0$  state is known as the *singlet* state.

Clebsch-Gordon coefficients for adding spin one-half to spin one-half. Only non-zero coefficients are shown.

	$-1/2, -1/2$	$-1/2, 1/2$	$1/2, -1/2$	$1/2, 1/2$	$m_{s_1}, m_{s_2}$
[0.5ex] 1, -1	1				
[0.5ex] 1, 0		$1/\sqrt{2}$	$1/\sqrt{2}$		
[0.5ex] 0, 0		$1/\sqrt{2}$	$-1/\sqrt{2}$		
[0.5ex] 1, 1				1	
$s, m_s$					

The Clebsch-Gordon coefficients for adding spin one-half to spin one-half can easily be inferred from Table [t2] (with  $l = 1/2$ ), and are listed in Table [t4]. It follows from this table that the three triplet states are:

$$\begin{aligned}\chi_{1,-1}^{(2)} &= \chi_{-1/2,-1/2}^{(1)}, \\ \chi_{1,0}^{(2)} &= \frac{1}{\sqrt{2}} \left( \chi_{-1/2,1/2}^{(1)} + \chi_{1/2,-1/2}^{(1)} \right), \\ \chi_{1,1}^{(2)} &= \chi_{1/2,1/2}^{(1)},\end{aligned}$$

where  $\chi_{s,m_s}^{(2)}$  is shorthand for  $\chi_{s_1,s_2;s,m_s}^{(2)}$ , et cetera. Likewise, the singlet state is written:

$$\chi_{0,0}^{(2)} = \frac{1}{\sqrt{2}} \left( \chi_{-1/2,1/2}^{(1)} - \chi_{1/2,-1/2}^{(1)} \right). \quad (10.3.3)$$

## Contributors and Attributions

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