

12.9: Radiation from Harmonic Oscillator

Consider an electron in a one-dimensional harmonic oscillator potential aligned along the x -axis. According to Section [sosc], the unperturbed energy eigenvalues of the system are

$$E_n = (n + 1/2) \hbar \omega_0, \quad (12.9.1)$$

where ω_0 is the frequency of the corresponding classical oscillator. Here, the quantum number n takes the values $0, 1, 2, \dots$. Let the $\psi_n(x)$ be the (real) properly normalized unperturbed eigenstates of the system.

Suppose that the electron is initially in an excited state: that is, $n > 0$. In principle, the electron can decay to a lower energy state via the spontaneous emission of a photon of the appropriate frequency. Let us investigate this effect. Now, according to Equation ([e3.115]), the system can only make a spontaneous transition from an energy state corresponding to the quantum number n to one corresponding to the quantum number n' if the associated electric dipole moment

$$(d_x)_{n,n'} = \langle n | e x | n' \rangle = e \int_{-\infty}^{\infty} \psi_n(x) x \psi_{n'}(x) dx \quad (12.9.2)$$

is non-zero [because $d_{if} \equiv (d_x)_{n,n'}$ for the case in hand]. However, according to Equation ([e5.xxx]),

$$\int_{-\infty}^{\infty} \psi_n x \psi_{n'} dx = \sqrt{\frac{\hbar}{2 m_e \omega_0}} \left(\sqrt{n} \delta_{n,n'+1} + \sqrt{n'} \delta_{n,n'-1} \right). \quad (12.9.3)$$

Because we are dealing with emission, we must have $n > n'$. Hence, we obtain

$$(d_x)_{n,n'} = e \sqrt{\frac{\hbar n}{2 m_e \omega_0}} \delta_{n,n'+1}. \quad (12.9.4)$$

It is clear that (in the electric dipole approximation) we can only have spontaneous emission between states whose quantum numbers differ by unity. Thus, the frequency of the photon emitted when the n th excited state decays is

$$\omega_{n,n-1} = \frac{E_n - E_{n-1}}{\hbar} = \omega_0. \quad (12.9.5)$$

Hence, we conclude that, no matter which state decays, the emitted photon always has the same frequency as the classical oscillator.

According to Equation ([e3.115]), the decay rate of the n th excited state is given by

$$w_n = \frac{\omega_{n,n-1}^3 (d_x)_{n,n-1}^2}{3\pi \epsilon_0 \hbar c^3}. \quad (12.9.6)$$

It follows that

$$w_n = \frac{n e^2 \omega_0^2}{6\pi \epsilon_0 m_e c^3}. \quad (12.9.7)$$

The mean radiated power is simply

$$P_n = \hbar \omega_0 w_n = \frac{e^2 \omega_0^2}{6\pi \epsilon_0 m_e c^3} [E_n - (1/2) \hbar \omega_0]. \quad (12.9.8)$$

Classically, an electron in a one-dimensional oscillator potential radiates at the oscillation frequency ω_0 with the mean power

$$P = \frac{e^2 \omega_0^2}{6\pi \epsilon_0 m_e c^3} E, \quad (12.9.9)$$

where E is the oscillator energy. It can be seen that a quantum oscillator radiates in an almost exactly analogous manner to the equivalent classical oscillator. The only difference is the factor $(1/2) \hbar \omega_0$ in Equation ([e13.126])—this is needed to ensure that the ground-state of the quantum oscillator does not radiate.

Contributors and Attributions

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