

2.6: Wien's Law

The wavelengths or frequencies at which these functions reach a maximum, and what these maximum values are, can be found by differentiation of these functions. They do not all come to a maximum at the same wavelength. For the four Planck functions discussed in Section 2.6 (Equations 2.6.1- 2.6.4), the wavelengths or frequencies at which the maxima occur are given by:

For Equation 2.6.1:

$$\lambda = W_1/T \quad (2.6.1)$$

For Equation 2.6.2:

$$\lambda = W_2/T \quad (2.6.2)$$

For Equation 2.6.3:

$$\nu = W_3 T \quad (2.6.3)$$

For Equation 2.6.4:

$$\nu = W_4 T \quad (2.6.4)$$

Any of these equations (but more usually the first one) may be referred to as *Wien's law*.

The constants are

$$W_n = \frac{hc}{kx_n}, \quad (n = 1, 2) \quad (2.6.5)$$

$$W_n = \frac{kx_n}{h}, \quad (n = 3, 4) \quad (2.6.6)$$

where the x_n are the solutions of

$$x_n = (6 - n) (1 - e^{-x_n}) \quad (2.6.7)$$

and have the values

$$x_1 = 4.965114 \quad (2.6.8)$$

$$x_2 = 3.920690 \quad (2.6.9)$$

$$x_3 = 2.821439 \quad (2.6.10)$$

$$x_4 = 1.593624 \quad (2.6.11)$$

The Wien constants then have the values

$$W_1 = 2.8978 \times 10^{-3} \text{ m K} \quad (2.6.12)$$

$$W_2 = 3.6697 \times 10^{-3} \text{ m K} \quad (2.6.13)$$

$$W_3 = 5.8790 \times 10^{10} \text{ Hz K}^{-1} \quad (2.6.14)$$

$$W_4 = 3.3206 \times 10^{10} \text{ Hz K}^{-1} \quad (2.6.15)$$

The maximum ordinates of the functions are given by

$$M_\lambda(\text{max}) = A_1 T^5 \quad (2.6.16)$$

$$N_\lambda(\text{max}) = A_2 T^4 \quad (2.6.17)$$

$$M_\nu(\text{max}) = A_3 T^3 \quad (2.6.18)$$

$$N_\nu(\text{max}) = A_4 T^2 \quad (2.6.19)$$

The constants A_n are given by

$$A_n = \frac{2\pi k^{6-n} y_n}{h^4 c^3}, \quad (n = 1, 2) \quad (2.6.20)$$

$$A_n = \frac{2\pi k^{6-n} y_n}{h^2 c^2}, \quad (n = 3, 4) \quad (2.6.21)$$

where the y_n are dimensionless numbers defined by

$$y_n = \frac{x_n^{6-n}}{e^{x_n} - 1} \quad (2.6.22)$$

That is,

$$y_1 = 21.20144 \quad (2.6.23)$$

$$y_2 = 4.779841 \quad (2.6.24)$$

$$y_3 = 1.421435 \quad (2.6.25)$$

$$y_4 = 0.6476102 \quad (2.6.26)$$

The constants A_n therefore have the values

$$A_1 = 1.2867 \times 10^{-5} \text{ W m}^{-2} \text{ K}^{-5} \text{ m}^{-1} \quad (2.6.27)$$

$$A_2 = 2.1011 \times 10^{17} \text{ ph s}^{-1} \text{ m}^{-2} \text{ K}^{-4} \text{ m}^{-1} \quad (2.6.28)$$

$$A_3 = 5.9568 \times 10^{-19} \text{ W m}^{-2} \text{ K}^{-3} \text{ Hz}^{-1} \quad (2.6.29)$$

$$A_4 = 1.9657 \times 10^4 \text{ ph s}^{-1} \text{ m}^{-2} \text{ K}^{-2} \text{ Hz}^{-1} \quad (2.6.30)$$

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