

## 8.4: Conservation of Energy

Today, physics is pretty much founded on the belief that the energy of a closed system (defined as one that does not exchange energy with its surroundings—more on this in a minute) is always *conserved*: that is, internal processes and interactions will only cause energy to be “converted” from one form into another, but the total, after all the forms of energy available to the system have been carefully accounted for, will not change. This belief is based on countless experiments, on the one hand, and, on the other, on the fact that all the fundamental interactions that we are aware of do conserve a system’s total energy.

Of course, recognizing whether a system is “closed” or not depends on having first a complete catalogue of all the ways in which energy can be stored and exchanged—to make sure that there is, in fact, no exchange of energy going on with the surroundings. Note, incidentally, that a “closed” system is not necessarily the same thing as an “isolated” system: the former relates to the total energy, the latter to the total momentum. A parked car getting hotter in the sun is not a closed system (it is absorbing energy all the time) but, as far as its total momentum is concerned, it is certainly fair to call it “isolated.” (And as you keep this in mind, make sure you also do not mistake “isolated” for “insulated”!) Hopefully all these concepts will be further clarified when we introduce the additional auxiliary concepts of force, work, and heat (although the latter will not come until the end of the semester).

For a closed system, we can state the principle of conservation of energy (somewhat symbolically) in the form

$$K + U + E_{\text{source}} + E_{\text{diss}} = \text{constant} \quad (8.4.1)$$

where  $K$  is the total, macroscopic, kinetic energy;  $U$  the sum of all the applicable potential energies associated with the system’s *internal* interactions;  $E_{\text{source}}$  is any kind of internal energy (such as chemical energy) that is *not* described by a potential energy function, but can increase the system’s mechanical energy; and  $E_{\text{diss}}$  stands for the contents of the “dissipated energy reservoir”—typically thermal energy. As with the potential energy  $U$ , the absolute value of  $E_{\text{source}}$  and  $E_{\text{diss}}$  does not (usually) really matter: all we are interested in is how much they change in the course of the process under consideration.

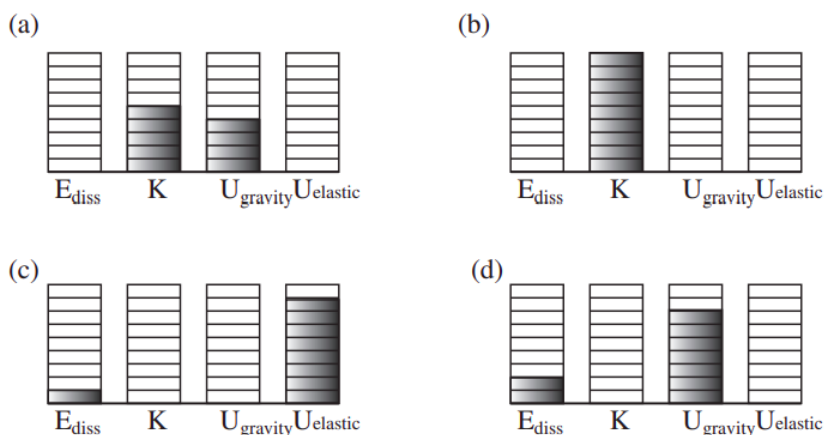


Figure 8.4.1: Energy bar diagrams for a system formed by the earth and a ball thrown downwards. (a) As the ball leaves the hand. (b) Just before it hits the ground. (c) During the collision, at the time of maximum compression. (d) At the top of the first bounce. The total number of energy “units” is the same in all the diagrams, as required by the principle of conservation of energy. From the diagrams you can tell that the coefficient of restitution  $e = \sqrt{7/9}$ .

Figure 8.4.1 above is an example of this kind of “energy accounting” for a ball bouncing on the ground. If the ball is thrown down, the system formed by the ball and the earth initially has both gravitational potential energy, and kinetic energy (diagram (a)). Note that we would normally need to worry about the Earth moving as well, but since the Earth is so massive, its movement is essentially zero.

As the ball falls, gravitational potential energy is being converted into kinetic energy, and the ball speeds up. As it is about to hit the ground (diagram (b)), the potential energy is zero and the kinetic energy is maximum. During the collision with the ground, all the kinetic energy is temporarily converted into other forms of energy, which are essentially elastic energy of deformation (like the energy in a spring) and some thermal energy (diagram (c)). When it bounces back, its kinetic energy will only be a fraction  $e^2$  of what it had before the collision (where  $e$  is the coefficient of restitution). This kinetic energy is all converted into gravitational potential energy as the ball reaches the top of its bounce (diagram (d)). Note there is more dissipated energy in diagram (d) than in

(c); this is because I have assumed that dissipation of energy takes place both during the compression and the subsequent expansion of the ball.

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